

The Structure of P_2I_4

By S. G. FRANKISS†*, F. A. MILLER†, H. STAMMREICH‡, and TH. TEIXEIRA SANS‡

(†Mellon Institute, Pittsburgh, Pennsylvania; ‡Department of Physics, University of São Paulo, São Paulo, Brazil; *Department of Chemistry, University College, London)

X-RAY-DIFFRACTION data have shown¹ that crystalline P_2I_4 has a *trans*- I_2PPI_2 structure with C_{2h} symmetry. In contrast, P_2I_4 dissolved in CS_2 has been reported to have a *gauche*-structure with C_2 symmetry from a determination of its dipole moment² and from its infrared spectrum³ between 300 and 400 cm^{-1} . From this it has been concluded⁴ that isolated P_2I_4 probably has a *gauche*-structure but that the effects of crystal packing may give rise to the *trans*-structure in the solid.

We have examined the infrared spectrum from 33 to 550 cm^{-1} and the Raman spectrum of P_2I_4 as a solid and as solutions in CS_2 , C_6H_6 , CH_2Cl_2 , CCl_4 , and CH_3I . The spectra are fully consistent only with P_2I_4 having the *trans*-structure in both states. This conclusion is based on the absence of coincidences between Raman and infrared bands, the observation of only two infrared-active and three Raman-active fundamentals (two of them

polarized and one depolarized) between 300 and 450 cm^{-1} , the similarity between the spectra of the solid and the solution, and the generally satisfactory vibrational assignments that can be made using the *trans*- I_2PPI_2 model.

The fundamental frequencies that are suggested for the solid are: a_g — 319, 307, 117, 78; a_u — 330, 89, 52; b_g — 328, 92, b_u — 301, 112, and 62 cm^{-1} . The probable fundamental frequencies for P_2I_4 in solution are: a_g — 316, 303, 114, 78; a_u — 327, 90, 51; b_g — 330, 95; b_u — 313, 109, and 65 cm^{-1} . Further details of these assignments will be presented separately.⁵

We conclude that P_2I_4 in the solid and liquid states has a geometry that is similar to the *trans*-structure proposed⁶ for P_2Cl_4 in its various states, but it is different from the *gauche*-structures that have been reported for P_2H_4 ,^{7,8} N_2H_4 ,⁹ and N_2F_4 .¹⁰⁻¹²

(Received, April 28th, 1966; Com. 289.)

¹ Y. C. Leung and J. Waser, *J. Phys. Chem.*, 1956, **60**, 539.

² M. Baudler and G. Fricke, *Z. anorg. Chem.*, 1963, **320**, 11.

³ A. H. Cowley and S. T. Cohen, *Inorg. Chem.*, 1965, **4**, 1200.

⁴ A. H. Cowley, *Chem. Rev.*, 1965, **65**, 617.

⁵ H. Stammreich, Th. Teixeira Sans, S. G. Frankiss, and F. A. Miller, *Spectrochim. Acta*, 1966, **22**, in the press.

⁶ S. G. Frankiss and F. A. Miller, *Spectrochim. Acta*, 1965, **21**, 1235.

⁷ E. R. Nixon, *J. Phys. Chem.*, 1956, **60**, 1054.

⁸ M. Baudler and L. Schmidt, *Z. anorg. Chem.*, 1957, **289**, 219; *Naturwiss.*, 1957, **44**, 488.

⁹ A. Yamaguchi, I. Ichishima, T. Shimanouchi, and S. Mizushima, *Spectrochim. Acta*, 1960, **16**, 1471.

¹⁰ D. R. Lide, Jr., and D. E. Mann, *J. Chem. Phys.*, 1959, **31**, 1129.

¹¹ J. R. Durig and R. C. Lord, *Spectrochim. Acta*, 1963, **19**, 1877.

¹² C. B. Colburn, F. A. Johnson and C. Haney, *J. Chem. Phys.*, 1965, **43**, 4526.